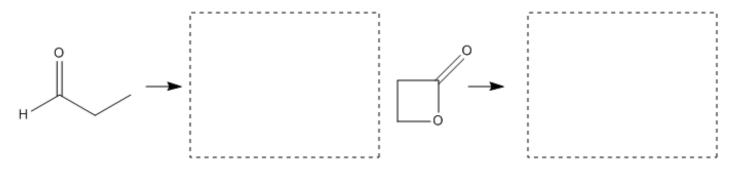
CONCEPT: REDUCING AGENT

- □ Reducing agents are used to reduce molecules
 - These reagents add hydrogens across π-bonds. Unsaturated hydrocarbons and carbonyls can be reduced.

General Mechanism: Nucleophilic addition of hydrogen

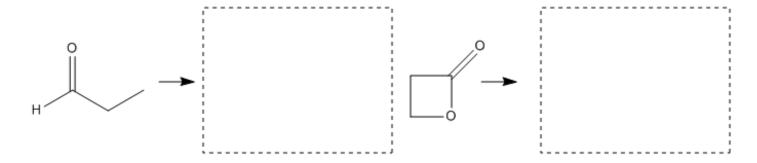
- \Box Reducing agents add to ALL π -bonds present. Multiple equivalents of hydrogen will react if possible.
 - LiAlH₄ is a **strong** reducing agent. It is able to reduce ANY carbonyl compound into alcohol

EXAMPLE: Draw the products of the following molecules reacted with LiAlH₄



□ NaBH₄ is a *weak* reducing agent. It can only add ____ equivalent of hydrogen and can only reduce CHO and ketones

EXAMPLE: Draw the products of the following molecules reacted with NaBH₄



PRACTICE: Predict the product of the following reactions

