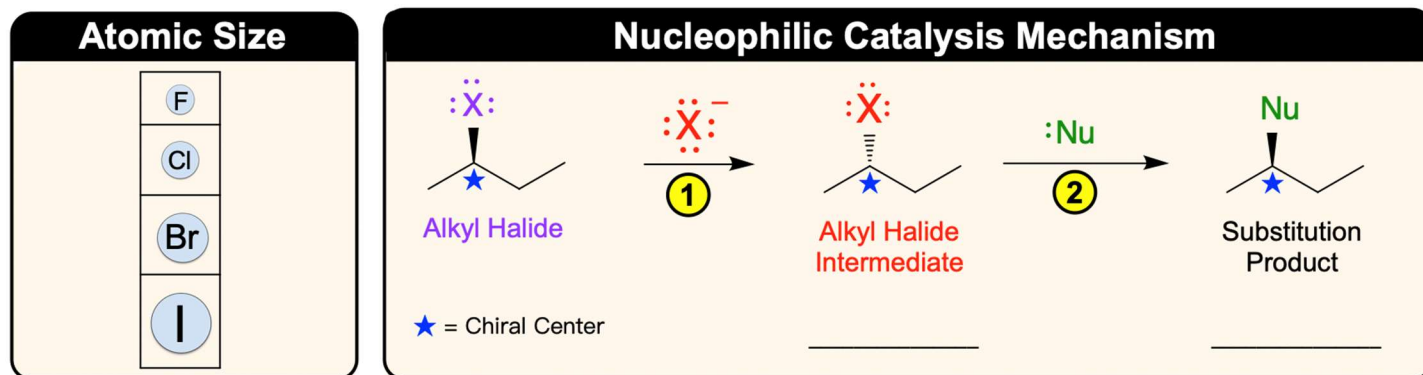
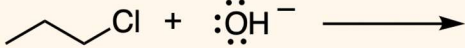
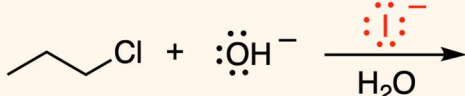


## CONCEPT: NUCLEOPHILIC CATALYSIS

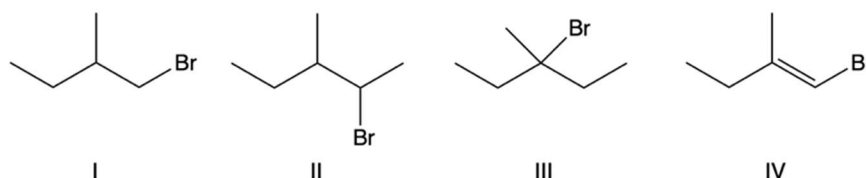
- In Orgo 1, we learned double  $S_N2$  caused \_\_\_\_\_ of R/S configurations, but it also serves as a *nucleophilic catalyst*.
  - Nucleophilic catalyst:** displaces leaving group with a \_\_\_\_\_ nucleophile/leaving group.



- Displacing the original halogen with a better nucleophile/leaving group also causes an \_\_\_\_ in the rate of reaction.

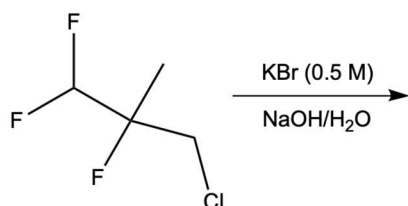
Uncatalyzed vs Catalyzed Reactions	
Reaction	Rate
<b>Uncatalyzed</b> 	$6.0 \times 10^{-6} \text{ s}^{-1}$
<b>Catalyzed</b> 	$3.7 \times 10^{-2} \text{ s}^{-1}$

**EXAMPLE:** Which of the following will undergo an  $S_N2$  reaction most readily when reacting with sodium hydroxide and a trace amount of iodide?

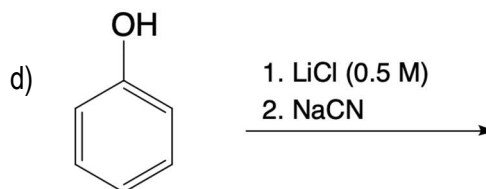
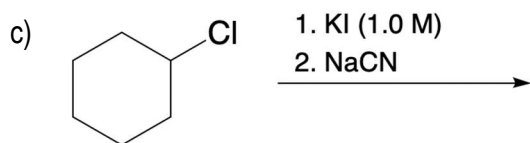
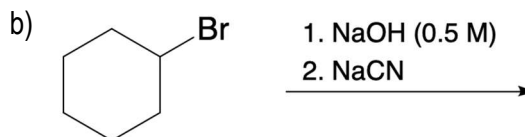
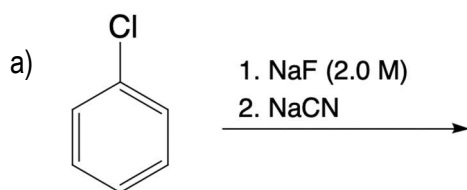


**CONCEPT: NUCLEOPHILIC CATALYSIS**

**PRACTICE:** Predict the final product from the chemical steps provided.



**PRACTICE:**



**PRACTICE:** Using 3-methyl-1-butene as a starting material, predict the final product based on the list of reagents given?

