## **CONCEPT: SOLVENTS**

Solvents are mostly inert compounds that provide a medium for the reaction to take place in.

□ Although extremely important in lab, they rarely affect the outcome of a written reaction in Orgo 1



## **Classification of Solvents**

<ul> <li>Protic solvents are solvents that display</li> </ul>	(stabilize	carbocations,	hinder	nucleophiles'	.)
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□ Therefore, we will prefer to run \_\_\_\_ & \_\_\_ in **protic** solvents, and \_\_\_\_ & \_\_\_ in **aprotic** solvents.

**EXAMPLE**: Identify the following solvents as apolar, polar aprotic or polar protic

	H_O_H	CH₃CN	CCl₄
Apolar Polar Aprotic Polar Protic			
	Ethanol	O N	H H
Apolar	Apolar	Apolar	Apolar
Polar Aprotic	Polar Aprotic	Polar Aprotic	☐ Polar Aprotic
☐ Polar Protic	☐ Polar Protic	☐ Polar Protic	☐ Polar Protic