## **CONCEPT:** NAMING CARBOXYLIC ACIDS

• Recall: Carboxylic acids possess a carbonyl carbon connected to a \_\_\_\_\_\_(OH) group.

• Set of rules for naming carboxylic acids are similar to aldehydes.

□ The carbonyl carbon of the carboxylic acid is always numbered \_\_\_\_.

□ Modify the ending from -\_\_\_\_ to -\_\_\_\_.

location-substituent-parent-modifier

**EXAMPLE**: Provide the systematic name for the following carboxylic acid.

$$\begin{array}{ccc} & \operatorname{Br} & \operatorname{O} \\ | & || \\ \operatorname{CH}_3\operatorname{CH}_2\operatorname{CH}\operatorname{CH}\operatorname{CH}_2\operatorname{C}\operatorname{-OH} \\ | & \\ \operatorname{CH}_2\operatorname{CH}_3 \end{array}$$

STEP 1: Find the longest carbon chain (parent chain) and assign name according to the prefixes and modifier.

□ Parent chain should include the COOH group and have \_\_\_\_\_ number of carbons.

 $\ \square$  If a tie between longest chains, choose chain with more substituents.

**STEP 2:** Assign name to all the substituents.

**STEP 3:** Start numbering the chain at the carbon of the \_\_\_\_\_ group.

STEP 4 to 6: Repeat steps from previous naming topics.

**PRACTICE:** If the substituent name of the OH group is hydroxy in the presence of a carboxylic acid, provide the systematic name for the following compound.