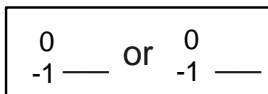


CONCEPT: BETA DECAY

- **Beta Decay** occurs when an unstable nucleus emits a _____ particle.

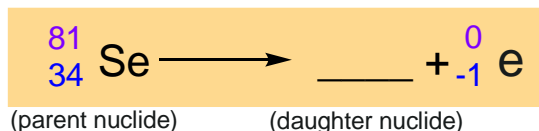
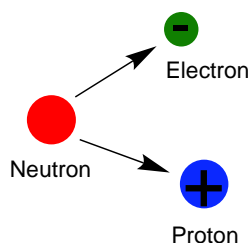
□ **Beta particle:** high energy and speed _____; symbolized as:



□ Usually occurs in nuclei with excess number of _____.

- _____ # of neutrons, _____ # of protons

- Neutron splits into a _____ and an electron; electron is ejected from the nucleus.



EXAMPLE: Write a balanced nuclear reaction for beta decay of I-129.

Characteristics of Beta Particles

- Beta particles are much _____ than alpha particles, have _____ ionizing power but _____ penetrating power.

Types of Radioactivity

Type	Particle	Example	Size	Ionizing Power	Penetrating Power	Shield
Alpha Decay	$\begin{matrix} 4 \\ 2 \end{matrix} \alpha$	$\begin{matrix} 171 \\ 78 \end{matrix} \text{ Pt} \longrightarrow \begin{matrix} 167 \\ 76 \end{matrix} \text{ Os} + \begin{matrix} 4 \\ 2 \end{matrix} \alpha$	Largest	Highest	Lowest	Clothing, skin, paper, air
Beta Decay	_____	$\begin{matrix} 81 \\ 34 \end{matrix} \text{ Se} \longrightarrow \begin{matrix} 81 \\ 35 \end{matrix} \text{ Br} + \begin{matrix} 0 \\ -1 \end{matrix} \beta$	_____	_____	_____	_____

EXAMPLE: Which are **not** the characteristics of beta decay?

- a) Beta particles are smaller in size but have higher ionizing power due to their speed.
- b) A high-energy, high-speed electron is ejected from a nucleus of an unstable atom.
- c) Due to higher penetrating power, beta particles can not be blocked by your skin.
- d) Beta particle carries a mass number of zero.